Quizlet

26 Multiple choice questions

1	carbohydrates that	t cannot be hydrolysed into	two or more simpler sugars,	and are known	as single sugar units
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- a. monomer
- b. electrode
- c. homologous series
- d. monosaccharides
- 2. atoms or molecules that are reactive as they have unpaired electrons
 - a. feedstock
 - b. fuel cell
 - c. free radicals
 - d. fermentation
- 3. the amount of heat generated when one mole of a substance is completely combusted
 - a. fractional distillation
 - b. fraction
 - c. fermentation
 - d. molar heat of combustion
- 4. the decomposition of carbohydrates by micro-organisms in the absence of air
 - a. feedstock
 - b. fermenting
 - c. fermentation
 - d. fraction
- 5. atoms of the same element that have the same atomic number but different mass number i.e. they have the same number of protons, but a different number of neutrons
 - a. enzymes
 - b. isomer
 - c. monomer
 - d. isotopes
- 6. a reaction that proceeds by absorbing heat
 - a. electrode
 - b. isomer
 - c. monomer
 - d. endothermic

7.	the number of protons and neutrons in the nucleus of an atom		
	a. isomer		
	b. f	fissionable	
	c. r	mass number	
	d. r	monomer	
8. protein molecules that act as catalysts; found in all organisms			
	a. i	somer	
	b. r	monomer	
	C. 6	enzymes	
	d. i	sotopes	
9. a compound that contains the same numbers and kinds of atoms as another, but has a different structure		pound that contains the same numbers and kinds of atoms as another, but has a different structure	
	a. i	sotopes	
	b. 6	enzymes	
	c. i	somer	
	d. r	monomer	
10.	a carb	on compound, also known as an organic compound, which is composed of only carbon and hydrogen atoms	
	a. i	somer	
	b. h	nydrocarbon	
	c. f	raction	
	d. 6	endothermic	
11.	a proc	ess by which substances are vaporised and then condensed by cooling	
	a. f	raction	
	b. f	functional group	
	c. f	ractional distillation	
	d. f	Fermentation	
12.	a famil	ly of similar carbon compounds with differing number of CH2 groups and containing the same functioning	
	a. r	mass number	
	b. r	monosaccharides	
	c. ł	nomologous series	
	d. r	monomer	

13.	a number of connected atoms whose presence in a molecule gives rise to characteristic chemical properties	
	a. fraction	
	b. electrolyte	
	c. fissionable	
	d. functional group	
14.	(C6H12O6) a naturally occurring six-carbon sugar	
	a. electrode	
	b. glucose	
	c. isotopes	
	d. isomer	
15.	a reactant used in an industrial process that is made from raw materials	
	a. fuel cell	
	b. electrode	
	c. feedstock	
	d. fermenting	
16.	a nucleus that is able to be split	
	a. isotopes	
	b. fissionable	
	c. mass number	
	d. fossil fuel	
17.	the small chained chemical unit that must be linked over and over to form a larger chain	
	a. isomer	
	b. monomer	
	c. isotopes	
	d. enzymes	
18.	an aqueous solution or molten substance that contains ions that are free to move and allows charges to move between electrodes	ž
	a. glucose	
	b. electrode	
	c. fraction	
	d. electrolyte	

19.	the breaking down of complex molecules brought about by some state of activity, such as when grape sugar is changed into ethanol by yeast enzymes				
	a. feedstock				
	b. fraction				
	c. fermenting				
	d. fermentation				
20.	an electrochemical cell in which the spontaneous occurrence of electrode reactions produces electrical energy a. mass number				
	b. glucose				
	c. fuel cell				
	d. galvanic cell				
21.	the system provided by the IUPAC for clearly naming chemicals with an explicit or implied relationship to the structur of compounds				
	a. fermentation				
	b. isomer				
	c. monomer				
	d. IUPAC nomenclature				
22.	a naturally occurring energy source formed within the Earth's crust from decomposing plant and animal matter				
	a. fissionable				
	b. fossil fuel				
	c. mass number				
	d. fuel cell				
23.	an electric cell that converts the chemical energy of a fuel, such as hydrogen, directly into electrical energy without the need to burn the fuel				
	a. fuel cell				
	b. feedstock				
	c. fossil fuel				
	d. glucose				
24.	a chemical compound separated from a large number of compounds within a mixture				
	a. fraction				
	b. fermentation				
	c. electrode				

d. fermenting

- 25. the potential voltage difference across the electrodes of a galvanic cell when a negligibly small current is being drawn
 - a. endothermic
 - b. electromotive force (emf)
 - c. electrolyte
 - d. electrode
- 26. a conductor, usually a metal or graphite, through which an electrical current can enter or leave a medium, such as a solution
 - a. fraction
 - b. electrode
 - c. electrolyte
 - d. glucose